

Target	=>	Synthon	Type of reaction	Reagents	Mechanism facts	Comments
alkoxide (RO ⁻)	=>	alcohol (ROH)	acid/base	STRONG base (e.g. NaNH ₂)		chap 3
alkynide	=>	alkyne	acid/base	STRONG base (e.g. NaNH ₂)		chap 3
carboxylate (RCOO ⁻)	=>	carboxylic acid	acid/base	NaOH or stronger		chap 3
alkane	=>	alkene	addition	H ₂ / Pt or Pd or Ni	syn-addition	chap 4 & 7
alkane	=>	alkyne	addition	H ₂ / Pt or Pd or Ni		chap 4 & 7
alkyne	=>	alkyl halide	substitution	alkynide	inversion of configuration; CHECK TABLE 6.6	chap 4, 6, & 7
alcohol	=>	alkyl halide	substitution	NaOH	check table 6.6; Inversion with SN ₂ , racemization with SN ₁ aprotic solvent helps	chap 6
ether	=>	alkyl halide	substitution	alkoxide	check table 6.6; Inversion with SN ₂ , racemization with SN ₁ aprotic solvent helps	chap 6
thiol	=>	alkyl halide	substitution	NaSH	check table 6.6; Inversion with SN ₂ , racemization with SN ₁ aprotic solvent helps	chap 6
thioether	=>	alkyl halide	substitution	NaSR	check table 6.6; Inversion with SN ₂ , racemization with SN ₁ aprotic solvent helps	chap 6
nitrile	=>	alkyl halide	substitution	NaCN	check table 6.6; Inversion with SN ₂ , racemization with SN ₁ aprotic solvent helps	chap 6
iodide	=>	alkyl halide	substitution	KI	check table 6.6; Inversion with SN ₂ , racemization with SN ₁ aprotic solvent helps	chap 6
carboxylic ester	=>	alkyl halide	substitution	carboxylate	check table 6.6; Inversion with SN ₂ , racemization with SN ₁ aprotic solvent helps	chap 6
alkene	=>	alkyl halide	elimination	KOH / ethanol / heat	check table 6.6; anti-periplanar; Zaitsev	chap 6 & 7
alkene	=>	alcohol	elimination	strong acid and heat	carbocation mechanism; Zaitsev; 3° easiest, 1° most difficult	chap 7
alkene, <i>cis</i> -	=>	alkyne	addition	H ₂ / P-2 or Lindlar's catalyst	syn addition	chap 7
alkyne	=>	vicinal dihalide	elimination	STRONG base (e.g. NaNH ₂)	two stages - first vinyl halide; then - with more difficulty - alkyne	chap 7
vinyl halide	=>	vicinal dihalide	elimination	KOH / ethanol / heat	see alkyne => vicinal dihalide	chap 7
alkyne	=>	vinyl halide	elimination	STRONG base (e.g. NaNH ₂)	see alkyne => vicinal dihalide	chap 7
alkyl halide	=>	alkene	addition	HX	carbocation mechanism; Markovnikov	chap 8
alcohol	=>	alkene	addition	H ₂ O; H ⁺ catalyst	carbocation mechanism; Markovnikov	chap 8
alcohol	=>	alkene	addition	H ₂ O / Hg(OAc) ₂ / THF 2) NaBH ₄ / base	no carbocation; Markovnikov	chap 8

alcohol	=>	alkene	addition	BH3 / THF 2) H2O2 / base	anti-Markovnikov; boron to least hindered position syn-addition	chap 8
vicinal dihalide	=>	alkene	addition	X2 / CCl4	anti-addition	chap 8
halohydrin	=>	alkene	addition	X2 / H2O	anti-addition	chap 8
cyclopropane	=>	alkene	cycloaddition	CH2N2 -or- CH2I2 / Zn(Cu)	stereospecific; CHCl3 / KOt-Bu for 1,1-dichlorocyclopropane	chap 8
vicinal diol	=>	alkene	addition	NMO / OsO4 catalyst; -or- OsO4 then NaHSO3; -or- cold dilute basic KMnO4	syn-addition	chap 8
carboxylic acid	=>	alkene (=CHR)	oxidative cleavage	KMnO4, base, heat; 2) H+		chap 8
ketone	=>	alkene (=CR2)	oxidative cleavage	KMnO4, base, heat -or- O3, CH2Cl2, -78 °C; 2) Me2S		chap 8
aldehyde	=>	alkene (=CHR)	oxidative cleavage	O3, CH2Cl2, -78 °C; 2) Me2S		chap 8
vinyl halide	=>	alkyne	addition	1 mol HX	Markovnikov	chap 8
geminal dihalide	=>	alkyne	addition	xs HX	Markovnikov	chap 8
geminal dihalide	=>	vinyl halide	addition	HX	Markovnikov	chap 8
<i>trans</i> vicinal dihaloalkene	=>	alkyne	addition	1 mol X2 / CCl4	anti addition	chap 8
$\alpha,\alpha,\beta,\beta$ -tetrahaloalkane	=>	alkyne	addition	xs X2 / CCl4		chap 8
carboxylic acid	=>	alkyne	oxidative cleavage	KMnO4, base, heat, 2) H+ -or- O3, CH2Cl2, -78 °C; 2) Me2S	triple bonded carbon must be non-terminal; otherwise gives CO2	chap 8
alkyl halide	=>	alkane	substitution	X2 / light or heat alt: NBS / light (Br only)	radical; reactivity: F >> Cl > Br (I unreactive) selectivity: 3° substitution > 2° > 1° > methyl Br is most selective; F is least selective	chap 10
alkyl halide	=>	alkene	addition	HBr / peroxide	radical; anti-Markovnikov addition of HBr	chap 10
polymer	=>	alkene	addition	peroxide -or- AIBN	radical chain reaction	chap 10